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MATERIAL SAFETY DATA SHEET

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Product Name: Industrial Bleach 6%

Synonyms: Sodium oxychloride; Soda bleach liquor; Javel water; Clorox; Javex.

Chemical Family: Hydrochlorous acid, sodium salt.

Application: Chemical intermediate. Laboratory reagent. Fungicide. Water-treatment. Pulp and paper. Bleaching

agent.

Distributed By: Chem-Wipe Industries Ltd.

24 - Hour Emergency Telephone Number (CANUTEC): (613) 996 6666 - Collect Calls Accepted

*666 Cellular Service

Preparation Date of MSDS: February 22/10
Prepared By: Chem-Wipe Industries Ltd.

2. COMPOSITION / INFORMATION ON INGREDENTS

Hazardous Components:

Ingredients Percent LD50s & LC50s Route & Species:

Sodium Hypochlorite, Solution
7.691, 52.0

Oral LD50 Mouse: 5800 mg/kg

7681-52-9 3-7 Ofai LD30 Mouse : 3800 Hig/s

3. HAZARDS IDENTIFICATION

Potential Acute Health Effects:

Eye Contact: Corrosive to eye tissue and may cause severe damage and blindness. This product causes corneal

scarring and clouding. Glaucoma. Cataracts.

Skin Contact: Corrosive. Causes severe burns. Prolonged and repeated exposure to dilute solutions often causes

irritation, redness, pain and drying and cracking of the skin. Toxic effects may be delayed. Avoid handling when the skin is moist, wet or abraided. May cause dermatitis, prolonged or repeated contact

may cause skin sensitization.

Inhalation: Corrosive to respiratory passages. Causes irritation of the mouth, nose and throat. Repeated and/or

prolonged exposures may cause productive cough, running nose, bronchopneumonia, pulmonary edema (fluid build-up in lungs) and reduction of pulmonary function. If mixed with acids or warmed to temperatures greater than 40 degrees Celsius, Sodium hypochlorite solutions release chlorine gas. This gas can cause severe irritation of the nose and throat. Exposures to high levels of chlorine gas

may result in severe lung damage.

Ingestion: Corrosive. Causes bums to the mouth, throat and stomach. Causes vomiting, nausea, and diarrhea.

Coma, shock and death may occur.

FIRST AID MEASURES

Wash eyes with water for a minimum of 30 minutes or until no evidence of the chemical remains. **Eye Contact:**

Hold eyelids open during flushing. Seek immediate medical attention.

Skin Contact: Remove contaminated clothing. Wash skin with water for at least 30 minutes, using soap if

available. Obtain medical attention immediately.

Inhalation: Remove person to fresh air. If not breathing, give artificial respiration. If breathing is difficult, get

immediate medical attention.

Ingestion: Rinse mouth with water. Do not induce vomiting. Do not give anything by mouth to an unconscious

person. If vomiting occurs spontaneously, keep head below hips to prevent aspiration of liquid into

the lungs. Seek immediate medical attention.

Due to the severely irritating or corrosive nature of the material, swallowing may lead to ulceration Notes to Physician:

and inflammation of the upper alimentary tract with hemorrhage and fluid loss. Also, perforation of

the esophagus or stomach may occur, leading to mediastinitis or peritonitis and the resultant

complications.

5. FIRE FIGHTING MEASURES

Flash Point (°C): None - will not burn.

Flash Point Method: Not applicable. **Auto Ignition Temperature (°C):** Not applicable. Flammable Limits in Air - Lower (%): Not applicable.

Flammable Limits in Air - Upper (%): Not applicable.

Extinguishing Media: Use extinguishing media appropriate for surrounding fire.

Keep containers cool to prevent rupture and release of material. Closed **Special Exposure Hazards:**

containers may explode in fire. Spilled material may cause floors and

contact surfaces to become slippery.

Special Protective Equipment: Fire fighters should wear full protective clothing, including self-contained

breathing equipment.

Not available. **HMIS Ratings for this product are:**

ACCIDENTAL RELEASE MEASURES

Absorb with an inert dry material and place in an appropriate waste disposal container. Spilled **Procedure for Clean Up:**

material may cause floors and contact surfaces to become slippery. Dike and contain land spills;

contain water spills by booming. Ventilate area.

Personal Precautionary Measures: Wear appropriate protective equipment.

Prevent from entering sewers, waterways or low areas. Prevent **Environmental Precautionary Measures:**

contamination of soil. Consult local authorities.

HANDLING AND STORAGE

Handling: Use good personal hygiene. Use appropriate personnel protective equipment. Use with adequate ventilation.

Containers which have been exposed to heat may be under internal pressure. These should be cooled and

carefully vented before opening. When diluting, add this product to water in small amounts to avoid spattering. Never add water to this material.

Storage:

Equipment for storage, handling or transpiration should not be made of: tin, copper and its alloys, nickel and its alloys and iron. Some metals accelerate the decomposition of Sodium Hypochlorite. Store below 29 Degrees Celsius. Do not freeze. Store in a cool, dry, well ventilated area, away from heat and ignition sources. Store away from organic chemicals, strong bases, metal powders, carbides, sulfides, and any readily oxidizable material. Keep away from direct sunlight. Storage area should be equipped with corrosion-resistant floors, sumps and should have controlled drainage to a recovery tank.

EXPOSURE CONTROLS / PERSONAL PROTECTION

Local exhaust ventilation as necessary to maintain exposures to within applicable **Engineering Controls:**

> limits. Make up air should always be supplied to balance air exhausted (either generally or locally). For personnel entry into confined spaces (i.e. bulk storage tanks) a proper confined space entry procedure must be followed including ventilation and testing of tank atmosphere. Ventilation should be corrosive proof.

Wear a Niosh approved full face piece respirator for acid gases or a self-contained **Respiratory Protection:**

breathing apparatus for air concentration levels up to 5 ppm. NIOSH approved supplied air respirator when airborne concentrations exceed exposure limits.

Gloves: Impervious gloves. Neoprene gloves. Nitrile gloves. Rubber gloves.

Skin Protection: Neoprene coated apron or chemical resistant clothing. Impervious boots.

Eyes: Chemical safety goggles and/or full face shield to protect eyes and face, if product is

handled such that it could be splashed into eyes. Do NOT wear contact lenses.

Other Personal Protection Data: Ensure that eyewash stations and safety showers are proximal to the work-station

location.

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Hazardous Components:

Ingredients Percent **Exposure limit - OSHA**

Sodium Hypochlorite, Solution

7681-52-9

Not available.

PHYSICAL AND CHEMICAL PROPERTIES

Physical State: Liquid

Green to yellow. Color:

Odor: Chlorine. pH: 11.5-13 **Specific Gravity:** 1.165

Boiling Point (°C): Decomposes.

Freezing Point (°C): -25

Vapor Pressure: 17.5 mmHg **Vapor Density:** Not Available. % Volatile by Volume: Not Available. **Evaporation Rate:** Not Available. **Solubility:** Soluble in water. VOCs (lbs/gallon): Not Available.
Viscosity: Not Available.
Molecular Weight: Not Available.

10. STABILITY AND REACTIVITY

Chemical Stability: Unstable.

Hazardous Polymerization: Will not occur.

Conditions to Avoid: Avoid excessive heat, open flames and all ignition sources.

Materials to Avoid: Strong oxidizers. Reducing agents. Ammonia.

Hazardous Decomposition Products: When heated to decomposition, it emits acrid smoke and irritating fumes.

Chlorine. Oxides of sodium. Oxygen. Sodium Hypochlorite solutions

decompose slowly. Decomposition is accelerated by heat (temperatures above

40 degrees Celsius) and light.

Additional Information: Hypochlorites may react with primary amines to form nitrogen trichloride

which explodes spontaneously in air. Hypochlorite bleach reacts with urea to form nitrogen trichloride which explodes spontaneously in air. Some metals accelerate the decomposition of Sodium Hypochlorite. Nickel. Copper. Tin.

Iron and its alloys. Manganese.

11. TOXICOLOGICAL INFORMATION

Principle Routes of Exposure

Ingestion: Corrosive. Causes burns to the mouth, throat and stomach. Causes vomiting, nausea, and diarrhea.

Coma, shock and death may occur.

Shin Contact: Corrosive. Causes severe burns. Prolonged and repeated exposure to dilute solutions often causes

irritation, redness, pain and drying and cracking of the skin. Toxic effects may be delayed. Avoid handling when the skin is moist, wet or abraided. May cause dermatitis, prolonged or repeated contact

may cause skin sensitization.

Inhalation: Corrosive to respiratory passages. Causes irritation of the mouth, nose and throat. Repeated and/or

prolonged exposures may cause productive cough, running nose, bronchopneumonia, pulmonary edema (fluid build-up in lungs) and reduction of pulmonary function. If mixed with acids or warmed to temperatures greater than 40 degrees Celsius, Sodium hypochlorite solutions release chlorine gas. This gas can cause severe irritation of the nose and throat. Exposures to high levels of chlorine gas

may result in severe lung damage.

Eye Contact: Corrosive to eye tissue and may cause severe damage and blindness. This product causes

corneal scarring and clouding. Glaucoma. Cataracts.

Other: Sodium hydrochlorite may cause skin sensitization or other allergic responses. Aspiration may cause

lung damage. Corrosive effects on the skin and eyes may be delayed, and damage may occur without

the sensation or onset of pain.

Acute Test of Product:

Acute Oral LD50: Not Available.

Acute Dermal LD50: Not Available.

Acute inhalation LC50: Not Available.

Carcinogenicity:

Hazardous Components:

Ingredients	Percent	IARC - Group 1 (carcinogenic to humans)	IARC - Group 2A (probably carcinogenic)	IARC - Group 2B (possibly carcinogenic)	IARC - Group 3 (not classified)	IARC - Group 4 (probably not carcinogenic)
Sodium Hypochlorite, Solution 7681-52-9	3-7	Not listed.	Not listed.	Not listed.	Not listed.	Not listed.

Hazardous Components:

Ingredients Percent **ACGIH 2000 - Carcinogens**

Sodium Hypochlorite, Solution

7681-52-9 Not Listed. 3-7

Carcinogenicity Comment: No additional information available.

Genotoxicity: Not Available. Not Available.

Reproductive / Developmental

Toxicity:

Mutagenicity:

Not Available.

Not Available.

Teratogenicity: **Embryo Toxicity:** Not Available.

12. ECOLGICAL INFORMATION

Ecotoxicological Information:

Hazardous Components:

Ingredients	Percent	Ecotoxicity - Fish Species Data	Acute Crustaceans Toxicity:	Ecotoxicity - Freshwater Algae Data
Sodium Hypochlorite, Solution 7681-52-9	3-7	Not Available.	Not Available.	Not Available.

Harmful to aquatic life at low concentrations. Toxicity is primarily associated with pH. Other Information:

13. DISPOSAL CONSIDERATIONS

Disposal of Waste Method: Disposal of all wastes must be done in accordance with municipal, provincial and federal

regulations.

Contaminaed Packaging: Empty containers retain product residue (liquid and/or vapor) and can be dangerous. Do not

> expose such containers to heat, flame, sparks, static electricity or other sources of ignition; they may explode. Do not dispose of package until thoroughly washed out. Dispose of container

according to national or local regulations.

14. TRANSPORTATION INFORMATION

TDG (Canada):

TDG Proper Shipping Hypochlorite Solution

Name:

Hazard Class: 8

UN Number: UN1791

Packaging Group: III

Note: No additional remark.

Marine Pollutant: No.

15. REGULATORY INFORMATION

Canadian DSL Inventory Status: Listed

WHMIS Hazardous Class: D2B TOXIC MATERIALS

E CORROSIVE MATERIAL

Symbols:





16. OTHER INFORMATION

Additional Information: This product has been classified in accordance with the hazard criteria of the Canadian Controlled

Products Regulations (CPR) and the MSDS contains all the information required by the CPR.

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